Multiphase Coexistence in an Ionic Liquid: Violation of Gibbs Phase Rule?

Various phases appeared in a single-component 1-decyl-3-methylimidazolium nitrate ($[C_{10}mim][NO_3]$) ionic liquid. Time-resolved synchrotron small- and wide-angle X-ray scattering (SWAXS) could distinguish the phase transitions depending upon the cooling rate. On the SWAXS patterns, low-Q peaks representing a few kinds of layered structures were decomposed. Multiphase coexistence was observed in $[C_{10}mim][NO_3]$ at the specific cooling rates (8–9 K/min). Ionic liquid crystal, hybrid-layered crystal, and hexagonal close-packed structure coexisted simultaneously. Multiphase coexistence was induced in the vicinity of a multiphase coexistence point.

A Gibbs phase rule describes the degree of freedom of a multiphase system under equilibrium conditions. A quadruple point was found even in a single-component system using the Monte Carlo simulations [1]. Recently, a generalized Gibbs phase rule was proposed with additional parameters [1, 2].

lonic liquids (ILs) consisting of a cation and an anion are characterized by nanoheterogeneous liquid structures [3]. The representative cation is 1-alkyl-3-methylimidazolium, [C_nmim]⁺, where *n* reveals the alky chain length. Various phases in [C_nmim] [X] appeared on the *n*-temperature (*T*) plot (Fig. 1) [4]. A multiphase coexistence point (MCP) locates in the vicinity of n = 10. Conformational polymorph of $[C_n mim][NO_3]$ (n = 4, 6, and 8) was examined at high pressure (HP), relating to HP multiple-glass transitions [5]. The stable conformers of [C₁₀mim]⁺ were evaluated using density fluctuation theory (DFT) (Fig. 2) [6]. The conformers of [C₁₀mim]⁺ in the crystal states were observed using Raman spectroscopy. Moreover, crystal structures of [C₁₀mim][NO₃] were determined at low temperature (LT) and HP using

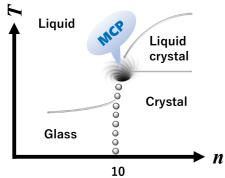


Figure 1: Phase diagram of $[C_n mim][X]$ on the n and temperature (7) scales.

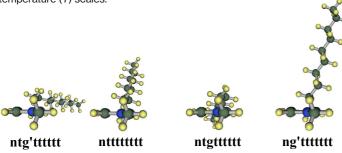


Figure 2: Optimized molecular conformations of [C₁₀mim]⁺ using DFT.

synchrotron small- and wide-angle X-ray scattering (SWAXS) [7]. A hybrid-layered structure with a lattice constant of 4.3 nm appeared at both LT and HP. [C₁₀mim][NO₃] was found to be flexible by pressure dependence of the lattice constants [7]. Here, we emphasize that the LT crystal polymorph was different from that at HP due to the packing polymorph. Cooling-rate dependent phase behaviors of [C₁₀mim][NO₃] were clarified using simultaneous SWAXS and differential scanning calorimetry (DSC) measurements [8]. Drastic phase changes were induced at a cooling rate of 8.1 K/min. Three broad exothermal peaks appeared on the DSC thermal trace at 8.1 K/min. Since the cooling rate was high, the phase changes were not distinguished using an *in-house* X-ray generator.

An in-situ observation is indispensable to extract hidden information about the complicated phase behaviors of [C₁₀mim][NO₃]. Synchrotron SWAXS with a high performance 2D detector (PILATUS) can obtain time-resolved 2D patterns even at the cooling rate of 20 K/min. For instance, the 2D SWAXS pattern of [C₁₀mim][NO₃] at 8 K/min is shown in Fig. 3. In the enlarged picture, the four Debye rings at low Q were observed distinctly. Generally, considering only one phase, a large unit cell is required to explain the four Bragg reflections at low Q. By changing the cooling rates systematically, the entirely different SWAXS patterns were detected at the minimum temperature, T_{min} (150 K) (Figs. 4(a)-4(f)). At 2 K/min, the hybridlayered crystal (C) phase appeared in the previous study (Fig. 4(a)) [7]. At 5 K/min, the ionic liquid crystal (ILC) phase existed partially in addition to the hybrid-

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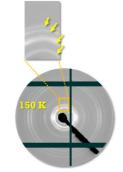


Figure 3: 2D SWAXS pattern of [C₁₀mim][NO₃] at cooling rate of 8 K/min.

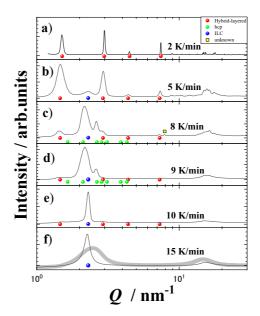


Figure 4: SWAXS patterns at 150 K with cooling rates of (a) 2, (b) 5, (c) 8, (d) 9, (e) 10, and (f) 15 K/min.

layered C phase (Fig. 4(b)). On the other hand, at 15 K/min (Fig. 4(f)), the ILC phase was preferred and the peak position was the same as the additional peak position at 5 K/min. More importantly, on the phase change from the hybrid-layered C at 2 K/min to the ILC at 15 K/min, a hexagonal close-packed (hcp) structure was induced at 8 and 9 K/min (Figs. 4(c) and 4(d)). Moreover, the amorphous halo appeared at 15 nm⁻¹. The sizes of the ILC, hybrid-layered C, and hcp phases are illustrated in Fig. 5. Since the temperature region of the ILC phase became narrower with increasing cooling rate [6, 8] (Fig. 1), the cooling path at 8 K/min could be approaching to the MCP. Here, we deduce that the specific fluctuations in the vicinity of the MCP caused the hcp phase at 8 K/min. The multiphase coexistence at 8 K/min is derived from the following three factors; (i) a flexible alkyl chain of [C₁₀mim]⁺, (ii) conformational polymorph of [C₁₀mim]⁺, and (iii) the specific fluctuations in the vicinity of the MCP. Based on three factors, we assume that alkyl chain entanglement of $[C_{10}mim]^+$ could be influenced extensively by the cooling rates. The appearance of well-ordered ILC or distorted ILC (d-ILC) phases implies that the chain entanglement/antiparallel packing contributed to the crystalline quality. In fact, the wellordered ILC appeared only at 8 and 9 K/min as seen in Fig. 6. Hence, in the vicinity of the MCP, the chain entanglement was suppressed at 8 and 9 K/min, and the hcp phase was formed additionally. Another significant point is that the reentrant phase transition of ILC was observed only at 8 and 9 K/min.

Time-resolved synchrotron SWAXS experiments can distinguish the cooling-rate effects for the complicated phase transitions of [C₁₀mim][NO₃]. Phase anomalies at 8 and 9 K/min in the vicinity of the MCP are summarized as follows; (i) multiphase coexistence

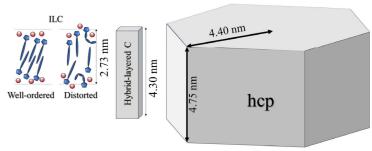


Figure 5: Sizes of the [C10mim]+ cation, well-ordered ILC, distorted ILC, hybrid-layered C, and hcp structures.

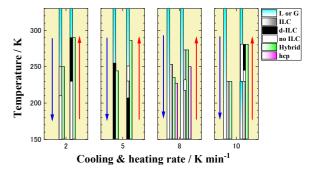


Figure 6: Cooling/heating rate dependence of the phase changes in [C₁₀mim][NO₃].

(additional hcp phase), (ii) the well-ordered ILC, and (iii) the reentrant phase transition of ILC. The well-ordered ILC, hybrid-layered C, and hcp phases coexisted with amorphous at T_{\min} . If amorphous is a liquid phase near the MCP, the four phases coexistence in a single-component system could be realized (Fig. 6) At the fixed pressure, the Gibbs phase rule is provided by F=C-N+1, where F, C, and N reveal the number of degrees of freedom, the number of different components, and the number of coexisting phases, respectively [2]. Violation of the Gibbs phase rule occurred in the single-component [C_{10} mim][NO₃]?

REFERENCES

- [1] K. Akahane, J. Russo and H. Tanaka, *Nature Commun.* 7, 12599 (2016).
- [2] V. F. D. Peters, M. Vis, Á. González García, H. H. Wensink and R. Tuinier, *Phys. Rev. Lett.* **125**, 127803 (2020).
- [3] J. N. A. Canongia Lopes and A. A. H. Pádua, *J. Phys. Chem. B* **110** .3330 (2006).
- [4] J. D. Holbrey and K. R. Seddon, Clean Prod. Proc. 1, 223 (1999).
- [5] H. Abe, T. Hirano, H. Kishimura, T. Takekiyo and Y. Yoshimura, J. Mol. Liq. 402, 124764 (2024).
- [6] H. Abe, S. Maruyama, H. Kishimura, M. Uruichi, D. Okuyama and H. Sagayama, J. Phys. Chem. Lett. 15, 10668 (2024).
- [7] H. Abe, Y. Yoshiichi, H. Kishimura and H. Sagayama, Chem. Phys. Lett. 827, 140685 (2023).
- [8] H. Abe and H. Kishimura, J. Mol. Lig. 352, 118695 (2022).

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6 HIGHLIGHTS